

Balancing Chemical Equations Worksheet

For each equation below, write the correct formula equation and then balance it.

1. phosphorus (P_4) + oxygen \rightarrow diphosphorus pentoxide



2. potassium hydroxide \rightarrow potassium oxide + water



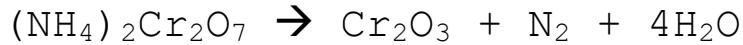
3. aluminum + copper(II) nitrate \rightarrow aluminum nitrate + Copper



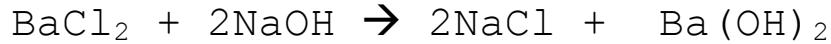
4. pentane (C_5H_{12}) + oxygen \rightarrow carbon dioxide + water



5. ammonium dichromate \rightarrow chromium(III) oxide + nitrogen + water



6. barium chloride + sodium hydroxide \rightarrow sodium chloride + barium hydroxide



7. nickel chlorate \rightarrow nickel chloride + oxygen



8. aluminum + hydrochloric acid \rightarrow aluminum chloride + Hydrogen



9. potassium hydroxide + antimony(III) chloride \rightarrow potassium chloride + antimony(III) oxide + water



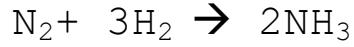
10. ethane (C_2H_6) + oxygen \rightarrow carbon dioxide + water



11. aluminum oxide \rightarrow aluminum + oxygen



12. nitrogen + hydrogen \rightarrow ammonia (NH_3)



13. calcium + water \rightarrow calcium hydroxide + hydrogen



14. propane (C_3H_8) + oxygen \rightarrow carbon dioxide + water



15. hydrogen + carbon monoxide + oxygen \rightarrow carbonic Acid



16. barium chlorate \rightarrow barium chloride + oxygen



17. glucose ($\text{C}_6\text{H}_{12}\text{O}_6$) + oxygen \rightarrow carbon dioxide +

Water



18. calcium chloride + sodium hydroxide \rightarrow sodium chloride + calcium hydroxide



19. methane (CH_4) + oxygen \rightarrow carbon dioxide + water

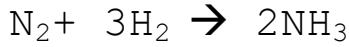


20. silver + nitric acid \rightarrow silver chloride + hydrogen

There is a typo chloride should be nitrate



21. nitrogen + hydrogen \rightarrow ammonia (NH_3)



22. sodium nitrate \rightarrow sodium nitrite + oxygen



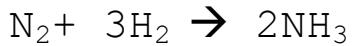
23. butane (C_4H_{10}) + oxygen \rightarrow carbon dioxide + water



24. aluminum oxide + hydrochloric acid \rightarrow aluminum chloride + water



25. nitrogen + hydrogen \rightarrow ammonia



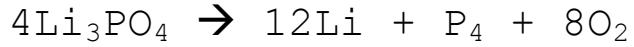
26. cesium iodide + bromine \rightarrow cesium bromide + iodine



27. ethane (C_2H_6) + oxygen \rightarrow carbon dioxide + water



28. lithium phosphate \rightarrow lithium + phosphorus (P_4) + Oxygen



29. aluminum + copper(II) iodide \rightarrow magnesium chlorate + copper

Typo, should be aluminum iodide



30. cobalt(III) chloride + potassium hydroxide \rightarrow potassium chloride + cobalt(III) hydroxide



31. manganese + hydrochloric acid \rightarrow manganese(II) chloride + hydrogen



32. aluminum sulfate + sodium hydrogen carbonate \rightarrow aluminum hydroxide + carbon dioxide + sodium sulfate

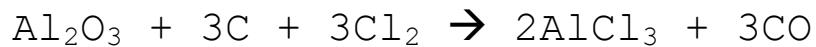


33. paraffin ($\text{C}_{25}\text{H}_{52}$) + oxygen \rightarrow carbon dioxide +

Water



34. aluminum oxide + carbon + chlorine \rightarrow aluminum chloride + carbon monoxide



35. magnesium + iron(III) nitrate \rightarrow magnesium nitrate + iron



36. strontium + chlorine + oxygen \rightarrow strontium chlorate
 $\text{Sr} + \text{Cl}_2 + 3\text{O}_2 \rightarrow \text{Sr}(\text{ClO}_3)_2$